

# Juxtamembrane Protein Segments that Contribute to Recruitment of Cholesterol into Domains<sup>†</sup>

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**ABSTRACT:** We investigated the properties of several peptides with sequences related to LWYIK, a segment found in the gp41 protein of HIV and believed to play a role in sequestering this protein to a cholesterol-rich domain in the membrane. This segment fulfills the requirements to be classified as a CRAC motif that has been suggested to predict those proteins that will partition into cholesterol-rich regions of the membrane. All of the peptides were studied with the terminal amino and carboxyl groups blocked, i.e., as *N*-acetyl-peptide-amides. Effects of cholesterol on the intensity of W emission generally parallel DSC evidence of sequestration of cholesterol. Modeling studies indicate that all of these peptides tend to partition with their mass center at the membrane interface at the level of the hydroxyl of cholesterol. Interaction with cholesterol is dual: van der Waals interactions between mainly hydrophobic surfaces and electrostatic stabilization of the cholesterol OH group. Thus, both experiments and modeling studies indicate that the preference of CRAC motifs for cholesterol-rich domains might be related to a membrane interfacial preference of the motif, to a capacity to wrap and block the cholesterol polar OH group by H-bond interactions, and to a capacity for peptide aromatic side chains to stack with cholesterol. These results were supported by studies of single mutations in the gp41 protein of HIV-1, in which L<sup>679</sup> is replaced with I. Despite the similarity of the properties of these amino acid residues, this single substitution resulted in a marked attenuation of the ability of JC53-BL HeLa-based HIV-1 indicator cells to form syncytia.

It has been demonstrated that peptides and proteins can cause the redistribution of cholesterol in membranes (1). Triggering domain formation may play an important role in signal transduction (2). One class of domains thought to be present in biological membranes has been termed “rafts”. Although there is controversy about the nature and prevalence of such domains, it is well established that caveolae are distinct domains in the membrane with many chemical and physical properties similar to those proposed for rafts.

Several features that facilitate the partitioning of proteins into raftlike domains have been identified. These features do not result in a specific stoichiometric binding between the protein and a specific lipid component, but rather a

preferential interaction of a protein with certain lipid components. Among the features that promote a more favorable partition into cholesterol-rich domains of membranes are certain types of protein lipidation (3). Some integral membrane proteins, devoid of lipidation, also partition into cholesterol-rich domains. Tryptophan at the flanking regions of these helices has been suggested to be important for this partitioning (4). In addition, several proteins that interact with cholesterol have an amino acid sequence in the juxtamembrane region conforming to the pattern -L/V-(X)-(1–5)-Y-(X)(1–5)-R/K-, in which (X)(1–5) represents between one and five residues of any amino acid (5). This motif, defined by a peptide segment -L/V-(X)(1–5)-Y-(X)-(1–5)-R/K-, has been termed the CRAC<sup>1</sup> motif, corresponding to the cholesterol recognition/interaction amino acid consensus.

One protein that has a CRAC motif adjacent to its transmembrane helix is the HIV-1 fusion protein, gp41. This segment contains the sequence LWYIK. The role of the short sequence LWYIK in binding to cholesterol has been

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<sup>1</sup> Abbreviations: CRAC, cholesterol recognition/interaction amino acid consensus; HIV, human immunodeficiency virus; PC, phosphatidylcholine; SOPC, 1-stearoyl-2-oleoylphosphatidylcholine; SUV, sonicated unilamellar vesicle; MLV, multilamellar vesicle; DSC, differential scanning calorimetry; *T*<sub>m</sub>, phase transition temperature;  $\Delta H$ , calorimetric enthalpy; MPER, membrane-proximal external region; rmsd, root mean square deviation.

demonstrated by assessing the binding of maltose binding protein fusion proteins to cholesterol–hemisuccinate agarose (6) and by MAS/NMR and DSC studies (7). A longer juxtamembrane region of gp41 is W-rich and contains the LWYIK sequence at the carboxyl-terminal end. It has been shown to promote membrane fusion in mutational studies of the intact viral protein (8) as well as with the use of a 20-amino acid synthetic peptide (9). The amino-terminal segment of this peptide is thought to facilitate oligomerization of gp41 (10). Cholesterol was found to be required for HIV infection (11–13) as well as for fusion promoted by the synthetic peptide (9). Depletion of cholesterol from HIV results in a loss of infectivity (14). Although the LWYIK segment of the gp41 protein of HIV has a CRAC motif and there is good evidence that this region of the protein interacts with cholesterol, the general requirements for the CRAC motif are too flexible to be accurate. For example, there are homologous segments of the W-rich, juxtamembrane region of gp41 proteins of SIV and HIV-2 that interact with cholesterol but do not fulfill the requirements of a CRAC motif (15). This study was designed to systematically vary certain positions of the LWYIK segment and to determine which of the molecular features of the peptide are important for its ability to sequester cholesterol.

## MATERIALS AND METHODS

**Materials.** The peptides are all *N*-acetyl-peptide-amides, i.e., blocked at the terminal amino and carboxyl groups. The peptides were synthesized by SynPep Corp. (Dublin, CA) and purified by HPLC to >95% purity. Phospholipids and cholesterol were purchased from Avanti Polar Lipids (Alabaster, AL).

**Preparation of Samples for DSC.** Phospholipid and cholesterol were codissolved in a chloroform/methanol mixture (2/1, v/v). For samples containing peptide, an aliquot of a solution of the peptide in methanol was added to the lipid solution in the chloroform/methanol mixture. The solvent was then evaporated under a stream of nitrogen with constant rotation of a test tube so that a uniform film of lipid would be deposited over the bottom third of the tube. Last traces of solvent were removed by placing the tube under high vacuum for at least 2 h. The lipid film was then hydrated with 20 mM PIPES, 1 mM EDTA, and 150 mM NaCl with 0.002% NaN<sub>3</sub> (pH 7.40) and suspended by intermittent vortexing and heating to 50 °C over a period of 2 min under argon.

**Differential Scanning Calorimetry (DSC).** Measurements were taken using a nano differential scanning calorimeter (Calorimetry Sciences Corp., American Fork, UT). The scan rate was 2 °C/min, and there was a delay of 5 min between sequential scans in a series to allow for thermal equilibration. The features of the design of this instrument have been described previously (16). DSC curves were analyzed by using the fitting program DA-2 provided by Microcal Inc. (Northampton, MA) and plotted with Origin, version 5.0.

**Tryptophan Fluorescence Studies.** The peptide was dissolved in methanol or in 10 mM Hepes buffer, 0.14 M NaCl, and 0.1 mM EDTA (pH 7.4) to a final concentration of 20  $\mu$ M. The purity of the peptide was determined by the absorbance at 280 nm using an extinction coefficient calculated from the amino acid content (17). Only for these

fluorescence studies were SUVs used to reduce contributions from light scattering. SUVs were made from lipid films dried from the chloroform/methanol mixture (2/1, v/v). Films were hydrated with Hepes buffer by vortexing and then sonicated. Peptide solutions (20  $\mu$ M) were then added to the partially clarified liposome preparation before the fluorescence was measured. The final lipid/peptide molar ratio was 10. The excitation wavelength was 280 nm, with a 4 nm bandwidth in excitation and 8 nm in emission. Polarizers, set to 90° in excitation and 0° in emission, were used to reduce stray light. Fluorescence emission spectra were measured using ultra-sensitive quartz mirror microcuvettes (N. L. Vekshin, Institute of Cell Biophysics, Pushchino, Russia). Spectra were recorded at 25 °C using an SLM Aminco Bowman AB-2 spectrofluorimeter. The spectra were corrected for instrumental factors, and a buffer blank was subtracted. Inner filter effect corrections were applied. When the fluorescence of the peptide was measured in the presence of lipid, the same concentration of SUVs without peptide was used as a blank.

**Membrane Partitioning Studies.** Films containing 10 mol % peptide and a SOPC/cholesterol mixture (1/1) were prepared by solvent evaporation and subsequently hydrated, as described for the preparation of samples for DSC. The final lipid concentration was 2.5 mg/mL in 20 mM PIPES, 0.14 M NaCl, and 1 mM EDTA (pH 7.4). The suspension was vortexed well and frozen and thawed two times between liquid nitrogen and ~50 °C to remove lipid still on the walls of the tube. The suspension was then incubated for 30 min at 38 °C. After further vortexing, the MLVs were transferred to 1 mL silanized polycarbonate tubes and centrifuged at 200000g (100 000 rpm) for 150 min at 25 °C in a Sorvall RC M120 centrifuge. A clear supernatant was collected in silanized glass tubes and the absorbance read by scanning between 350 and 250 nm in a Varian Cary 50Bio spectrophotometer, using the supernatant from a suspension of lipid without peptide as a blank. The absorbance at 280 nm, corrected for the blank and for light scattering, was used to calculate the amount of peptide remaining in the supernatant, compared with the total concentration of peptide determined by absorbance before the addition of lipid.

## MODELING

**Structure Properties.** A search for the most favorable conformation of each of the peptides was made by selecting the lowest-energy combinations of favorable dihedral angles (18–20). The blocked pentapeptides have five peptide bonds, each of which is allowed to have any of six different sets of dihedral angles, resulting in 6<sup>5</sup> or 7776 different structures. This stereoalphabet method was used previously to study the loop conformation for helix–helix interactions (21). The energies of peptide conformations are calculated by an all atom description of structures and the addition of van der Waals, electrostatic, internal, and external hydrophobicity energy terms. The van der Waals contribution was calculated using the 6-12 Lennard-Jones description of the energy of interactions between unbonded atoms (21). Coulomb's equation was used for electrostatic interactions between unbonded charged atoms with a dielectric “constant” sigmoidally varying from 1 to 80 with a distance between atoms and using FCPAC partial atomic charges (22). The intramolecular hydrophobicity contribution to stability was calculated using atomic Etr (energy of transfer) and the fractions of

atomic surface covered by the atom in interactions. Seven atomic types are used for Etr (20). When structures are calculated in water, an external hydrophobicity energy term where the solvent-accessible surface of atoms is calculated by the method of Shrake and Rupley with 162 points accounts for the contribution of solvent (23, 24). From the 7776 calculated structures, the 100 most stable are saved (50 in water and 50 in water-free medium).

**Membrane Insertion.** Insertion of all these structures into the membrane was tested with the IMPALA systematic program (25). The minimal restraint energy profile of insertion of each peptide in membrane was determined by inserting separately the 100 previously selected structures into the membrane. Each molecule was moved every 1 Å step across the bilayer ( $z$  axis) and was rotated in the  $z$ - $x$  plane (5000 positions over 360°) at each step. The insertion profile of one peptide in membrane was set as the combination of the minima at all steps; hence, the profile could account for a different structure and a different angle of insertion at each level of the membrane.

**Cholesterol-Peptide Interactions.** Determining the interactions of the most stable conformation of each peptide in membrane with cholesterol extended the analysis. The best IMPALA position was used for each peptide conformation (100 conformations per peptide) and the best position for the cholesterol in the membrane. The two molecules were tested in the calculation by varying the position of cholesterol in the vertical ( $z$  axis) and horizontal (around the CRAC peptide) directions as well as by self-rotation around the long axis of cholesterol. All together, 5 720 400 relative positions between each peptide conformation and sterol were tested to finally retain the conformation of maximal intermolecular interactions that was refined by angular dynamics (32). The best complex among all complexes formed was evaluated on the basis of having optimal (i.e., minimal) values of intra- and intermolecule energies.

## CELL-CELL FUSION

**Cell Culture.** JC53-BL HeLa-based HIV-1 indicator cells (26) that are highly susceptible to HIV-1 infection were kindly provided by Tranzyme Inc. (Birmingham, AL). These cells contain reporter genes, luciferase and  $\beta$ -galactosidase, that are expressed in the presence of Tat, under the influence of HIV-1 LTR. African green monkey kidney (Cos-1) cells were purchased from American Type Culture Collection. Cells were subcultured every 3–4 days by trypsinization and maintained in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum and antibiotics.

**Mutagenesis and Plasmid Vectors.** An NL4.3 (laboratory-adapted X4 strain) KpnI–BamHI envelope fragment was cloned into the pSP72 vector. Mutant L679I of the HIV-1 membrane-proximal external region (MPER) was performed using QuikChange (Stratagene, La Jolla, CA). The mutated envelope fragment was subsequently subcloned into an SV-40-based vector (pSRHS) that expresses Tat and Env (8).

**Cell-Cell Fusion Assays.** Cos-1 cells were transfected with pSRHS constructs expressing wild-type and mutant HIV-1 envelope using Fugene 6 transfection reagent (Roche Diagnostics Corp., Indianapolis, IN). Twenty hours post-transfection, they were mixed in a 5/1 ratio with JC53-BL cells and replated in 24-well plates. After 20 h, the cells were

microscopically examined for fusion by counting the blue syncytia and the number of nuclei/syncytium following addition of 5-bromo-4-chloro-3-indolyl  $\beta$ -D-galactopyranoside (X-gal; Promega, Madison, WI) as described by Kimpton and Emerman (27). Fusion was also evaluated by measuring luciferase activity (Luciferase Assay System, Promega), according to the manufacturer's protocol. All fusion assays were carried out in triplicate.

## RESULTS

**DSC.** DSC can be used to assess the miscibility of cholesterol and SOPC. With increasing mole fractions of cholesterol, the melting transition of SOPC will become broader and have a reduced transition enthalpy. If a peptide promotes the rearrangement of cholesterol into cholesterol-rich domains, the remaining SOPC-rich domains will exhibit a more cooperative (sharper) phase transition of higher enthalpy. This is most easily discerned in the cooling scans, because of the low transition temperature of this phospholipid. The phase behavior of SOPC with several mole fractions of cholesterol in the presence of *N*-acetyl-LWYIK-amide is known (7). In this work, we study several other related blocked pentapeptides. In this new group of peptides, we kept positions 2 and 4 of LWYIK as W and I, respectively, but have made substitutions in the other three positions. Some of the homologues fulfilled the requirements of the CRAC motif, and others did not.

A pair of peptides that differ only by a single OH group on Y are *N*-acetyl-LWFIK-amide and *N*-acetyl-LWYIK-amide. However, although *N*-acetyl-LWFIK-amide is not an example of a CRAC motif, it does have effects similar to those of *N*-acetyl-LWYIK-amide in redistributing cholesterol (Figure 1). In contrast, substitution of Y with another aromatic amino acid, W, resulted in much less sequestering of cholesterol (Figure 2). A third similar peptide that does not fulfill the CRAC formula is *N*-acetyl-IWYIK-amide, since an L or V residue is required in the first position. Unlike the non-CRAC peptide *N*-acetyl-LWFIK-amide, the peptide *N*-acetyl-IWYIK-amide is not effective in depleting a domain of cholesterol (Figure 3).

A peptide that requires cholesterol to bind to membranes should not greatly affect the phase transition of pure SOPC. In general, the peptides used in this study do not lower the enthalpy of the phase transition of SOPC (Table 1, pure SOPC columns). One peptide, *N*-acetyl-IWYIK-amide, however, does cause the enthalpy of SOPC to decrease.

It is well-known that cholesterol broadens and lowers the main transition enthalpy of PC (Table 1). Peptides that preferentially bind to cholesterol-rich domains in mixtures with PC will deplete cholesterol from a region of the membrane, resulting in a peptide-induced increase in the enthalpy of the phospholipid chain melting transition ( $I$ ). The peptide *N*-acetyl-LWYIK-amide is the only peptide of this group that is very effective in doing this at 30% cholesterol (Figure 4). When the amount of cholesterol in the membrane exceeds 30%, several peptides cause an increase in the enthalpy of the SOPC phase transition enthalpy (Table 1 and Figure 4). Hence, analysis of the DSC data supports the suggestion of a cholesterol-sequestering effect that is selective for LWYIK at 30% cholesterol and occurs with other peptides at higher cholesterol concentrations.

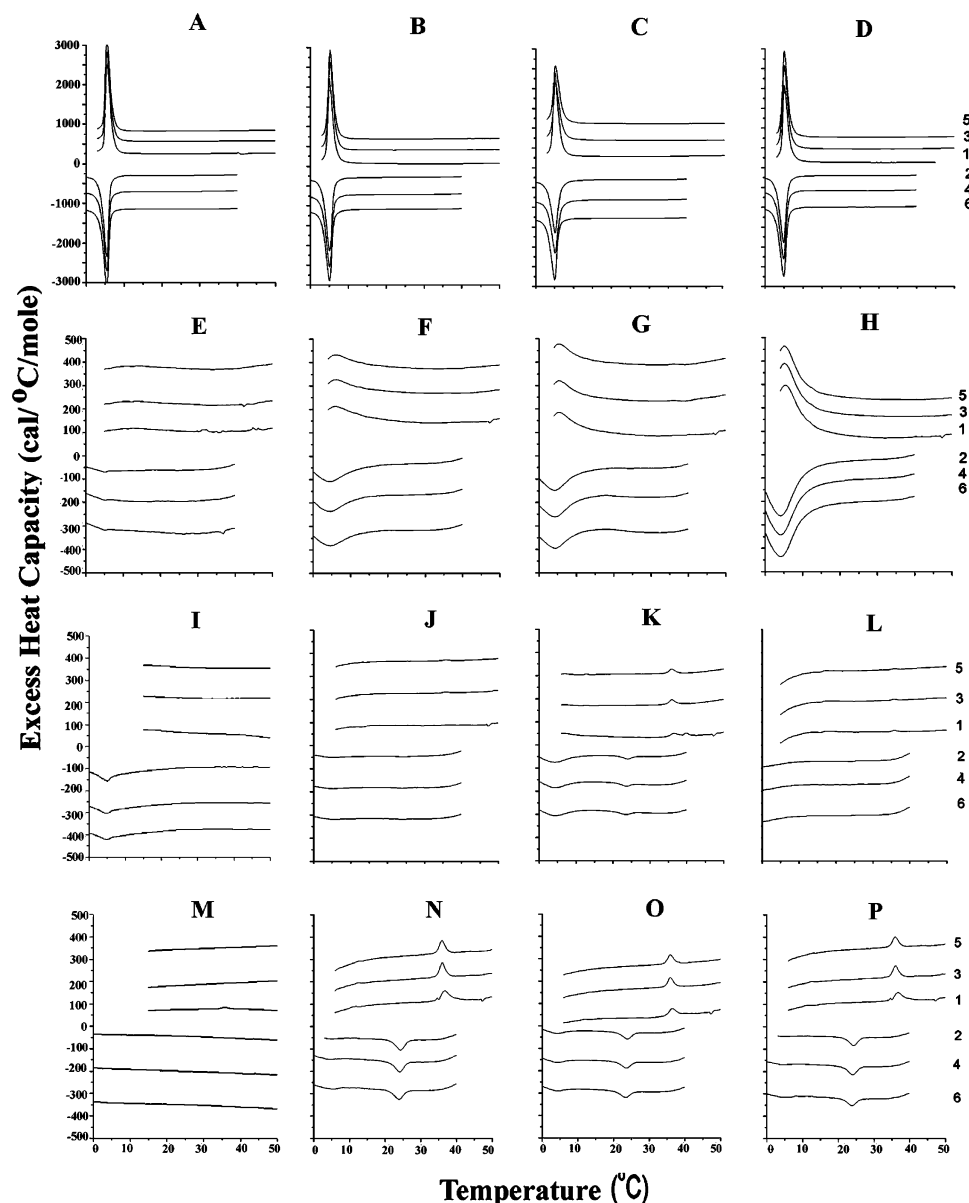


FIGURE 1: Differential scanning calorimetry of SOPC with 0 (A–D), 30 (E–H), 40 (I–L), and 50 mol % cholesterol (M–P) as well as 0 (A, E, I, and M), 5 (B, F, J, and N), 10 (C, G, K, and O), and 15 mol % *N*-acetyl-LWFIK-amide (D, H, L, and P). The scan rate was 2 K/min. The lipid concentration was 2.5 mg/mL in 20 mM PIPES, 1 mM EDTA, and 150 mM NaCl with 0.002% NaN<sub>3</sub> (pH 7.40). Sequential heating and cooling scans between 0 and 50 °C. Numbers are the order in which the scans were carried out, with scans 1, 3, and 5 being heating scans, each of which was followed by one of the cooling scans (2, 4, and 6). Scans were displaced along the y axis for clarity of presentation.

Another criterion for the formation of cholesterol-rich domains is that the sterol passes its solubility limit in a region of the membrane and can form crystallites. Anhydrous crystalline cholesterol is known to undergo a crystalline polymorphic phase transition at 38 °C on heating and at 23 °C on cooling at a scan rate of 2 °C/min (28), with an enthalpy of 910 cal/mol (29). At a SOPC/cholesterol molar ratio of 7/3, no cholesterol crystals, or at most traces of crystalline cholesterol, are found with any of the peptides. However, at a SOPC/cholesterol molar ratio of 6/4, anhydrous cholesterol crystals can be detected with all of the peptides, except for *N*-acetyl-LWYIH-amide (not shown). With a 1/1 mixture of SOPC and cholesterol, there is still no deposition of cholesterol in the absence of peptide, but all of the peptides used in this study cause some cholesterol crystals to form (Table 2). There are two mechanisms by which peptides can promote the formation of cholesterol-

rich domains: one by preferentially binding to these domains and stabilizing them and the other by being excluded from these domains and preferentially binding to cholesterol-depleted regions of the membrane, forcing cholesterol into another region of the membrane (1). From the formation of cholesterol crystallites, it is not possible to distinguish between these two mechanisms. In the case of peptides that are sequestered into cholesterol-rich domains, they will cause the formation of only cholesterol crystallites from the fraction of cholesterol to which the peptide is not directly bound. Cholesterol that is bound to peptide has a lower probability of forming crystals. Generally, the amount of cholesterol crystals formed does not increase proportionally to peptide concentration (Table 2). This is particularly striking in the case of *N*-acetyl-LWYIH-amide that has fewer cholesterol crystals with each successive increment of peptide concentration; this may be a consequence of more cholesterol being



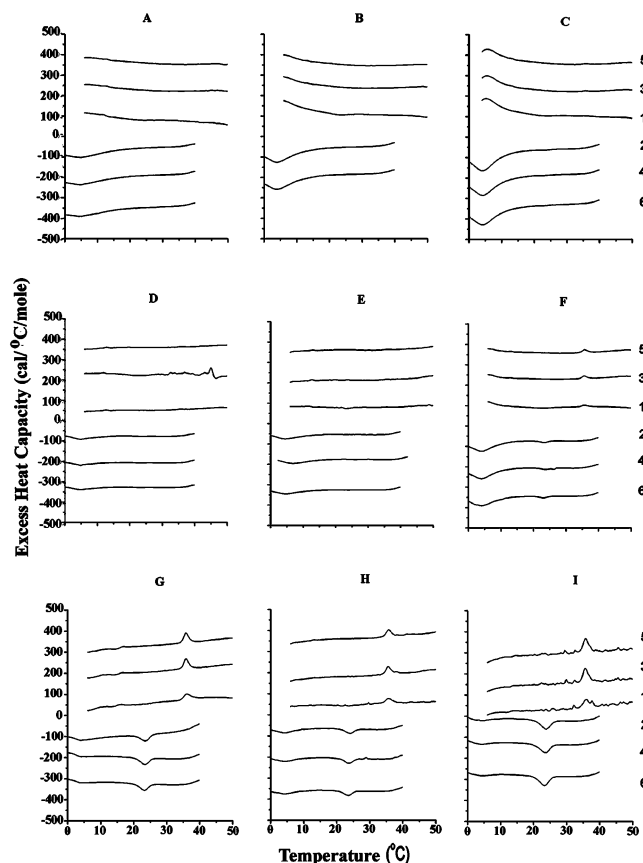


FIGURE 2: Differential scanning calorimetry of SOPC with 30 (A–C), 40 (D–F), and 50 mol % cholesterol (G–I) as well as 5 (A, D, and G), 10 (B, E, and H), and 15 mol % *N*-acetyl-LWWIK-amide (C, F, and I). For other details, see the legend of Figure 1.

directly bound to peptide as the peptide concentration increases.

**Tryptophan Fluorescence Studies.** Fluorescence emission spectra of each of the peptides were determined in a methanolic solution, in aqueous buffer, and in the presence of SUVs of SOPC or SOPC and cholesterol (1/1) at a lipid/peptide molar ratio of 10. There is little difference in the maximum emission wavelength among the seven peptides under any of the four conditions (Table 3).

**Membrane Partitioning.** The results of partitioning of *N*-acetyl-LWYIK-amide into SOPC/cholesterol SUVs (1/1) (Figure 5) are in reasonable agreement with our previously published results (7), particularly taking into account the differences in protocol in the two cases. However, there was some variation in the partitioning of other peptides into liposomes of SOPC and cholesterol (1/1) (Figure 5). In particular, *N*-acetyl-LWLIK-amide exhibited a lower level of membrane partitioning.

**Modeling.** Because of their small size, all these peptides have limited possibilities of stabilization via intramolecular interactions and hence might have a large structural diversity. This diversity will be restricted by the presence of bulky residues and by the proximity in the sequence of polar and hydrophobic residues. The rmsd of every structure was calculated on the basis of an all atom fit with the peptide conformation with the lower energy (the best structure). The values are ranked, and analysis of the plot of increasing values demonstrates either a stepped or a constant increase in the rmsd values (Figure 6). A constant variation of rmsd

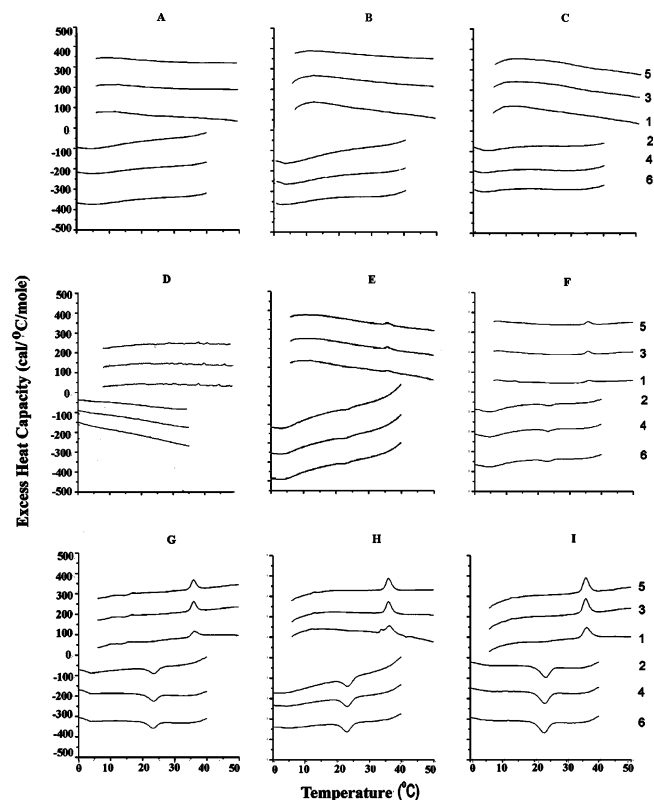


FIGURE 3: Differential scanning calorimetry of SOPC with 30 (A–C), 40 (D–F), and 50 mol % cholesterol (G–I) as well as 5 (A, D, and G), 10 (B, E, and H), and 15 mol % *N*-acetyl-IWFIK-amide (C, F, and I). For other details, see the legend of Figure 1.

Table 1: Enthalpies of the Chain Melting Transition of SOPC

| pure SOPC |                         | 7/3 SOPC/cholesterol |                         | 6/4 SOPC/cholesterol |                         |
|-----------|-------------------------|----------------------|-------------------------|----------------------|-------------------------|
| peptide   | $\Delta H$<br>(cal/mol) | peptide              | $\Delta H$<br>(cal/mol) | peptide              | $\Delta H$<br>(cal/mol) |
| none      | 3800                    | none                 | 520                     | none                 | 100                     |
| 5% LWYIK  | 3800                    | 5% LWYIK             | 1000                    | 5% LWYIK             | 100                     |
| 10% LWYIK | 3800                    | 10% LWYIK            | 1100                    | 10% LWYIK            | 300                     |
| 15% LWYIK | 3800                    | 15% LWYIK            | 1200                    | 15% LWYIK            | 500                     |
| 5% LWFIK  | 3750                    | 5% LWFIK             | 380                     | 5% LWFIK             | 65                      |
| 10% LWFIK | 3800                    | 10% LWFIK            | 440                     | 10% LWFIK            | 160                     |
| 15% LWFIK | 3700                    | 15% LWFIK            | 1300                    | 15% LWFIK            | broad                   |
| 5% LWYIR  | 4100                    | 5% LWYIR             | 550                     | 5% LWYIR             | 160                     |
| 10% LWYIR | 3600                    | 10% LWYIR            | 440                     | 10% LWYIR            | 250                     |
| 15% LWYIR | 3900                    | 15% LWYIR            | 440                     | 15% LWYIR            | 205                     |
| 5% LWYIH  | 3700                    | 5% LWYIH             | 520                     | 5% LWYIH             | 200                     |
| 10% LWYIH | 3900                    | 10% LWYIH            | 450                     | 10% LWYIH            | broad                   |
| 15% LWYIH | 3400                    | 15% LWYIH            | 325                     | 15% LWYIH            | broad                   |
| 5% LWWIK  | 3700                    | 5% LWWIK             | 148                     | 5% LWWIK             | ~0                      |
| 10% LWWIK | 3900                    | 10% LWWIK            | 245                     | 10% LWWIK            | 135                     |
| 15% LWWIK | 3700                    | 15% LWWIK            | 340                     | 15% LWWIK            | 195                     |
| 5% LWLIK  | 4000                    | 5% LWLIK             | 210                     | 5% LWLIK             | ~0                      |
| 10% LWLIK | 3900                    | 10% LWLIK            | 330                     | 10% LWLIK            | 110                     |
| 15% LWLIK | 3500                    | 15% LWLIK            | 280                     | 15% LWLIK            | 165                     |
| 5% IWYIK  | 3600                    | 5% IWYIK             | ~0                      | 5% IWYIK             | ~0                      |
| 10% IWYIK | 3200                    | 10% IWYIK            | ~0                      | 10% IWYIK            | 140                     |
| 15% IWYIK | 2600                    | 15% IWYIK            | ~0                      | 15% IWYIK            | 160                     |

(see LWFIK in Figure 6) indicates small discrete variations between structures, whereas a stepped increase (see LWWIK in Figure 6) indicates more distinct subpopulations. The number of structures within a 3.5 rmsd value difference is a subjective index that we used to evaluate this difference. For instance, changing L for I, two residues with the same number of carbons but an elongated instead of a bulky branched side chain, restricts structural diversity from the 55 conformations for LWYIK to only 27 conformations for

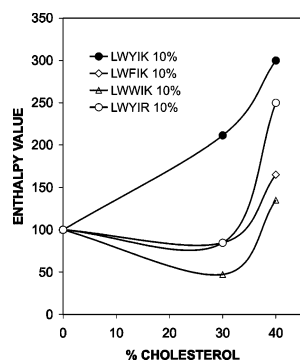


FIGURE 4: Graphic representation of the change in the SOPC phase transition enthalpy caused by the presence of selected peptides. The relative enthalpy value, defined as the enthalpy of the SOPC transition in the presence or absence of 10 mol % peptide times 100, is plotted as a function of the cholesterol content in the membrane: *N*-acetyl-LWYIK-amide (●), *N*-acetyl-LWFIK-amide (◇), *N*-acetyl-LWWIK-amide (△), and *N*-acetyl-LWYIR-amide (○).

Table 2: Enthalpies of the Polymorphic Transition of Anhydrous Cholesterol Present in Mixtures of Peptides with SOPC and Cholesterol (1/1)

| peptide   | $\Delta H$ (cal/mol) | peptide   | $\Delta H$ (cal/mol) |
|-----------|----------------------|-----------|----------------------|
| 5% LWYIK  | 25                   | 15% LWYIH | 60                   |
| 10% LWYIK | 80                   | 5% LWWIK  | 125                  |
| 15% LWYIK | 60                   | 10% LWWIK | 80                   |
| 5% LWFIK  | 120                  | 15% LWWIK | 140                  |
| 10% LWFIK | 125                  | 5% LWLIK  | 60                   |
| 15% LWFIK | 170                  | 10% LWLIK | 65                   |
| 5% LWYIR  | 20                   | 15% LWLIK | 120                  |
| 10% LWYIR | 70                   | 5% IWYIK  | 100                  |
| 15% LWYIR | 65                   | 10% IWYIK | 140                  |
| 5% LWYIH  | 180                  | 15% IWYIK | 175                  |
| 10% LWYIH | 90                   |           |                      |

IWYIK (Figure 6). The same is true when Y is changed to W (LWYIK compared to LWWIK), whereas changing Y to F (LWYIK compared to LWFIK) diversifies the panel of conformations, suggesting more rigidity of the LWYIK structure. Changing K to R (LWYIK compared to LWYIR) also rigidifies the structure most likely because R is much more polar than K and tends to point away from the rest of the peptide. This analysis suggests the following rough classification of increasing structure flexibility: LWYIR < LWWIK < IWYIK < LWYIH < LWYIK < LWFIK.

Calculating the energy for insertion for the 100 most stable conformations of each of these peptides (50 structures calculated in water and 50 calculated in a hydrophobic environment) supports the conclusion that all are more stable in a membrane than in water because all peptides have minimal restraint energy in a membrane (Figure 7). Moreover, all favor partitioning in the lipid acyl chain–polar interface. Some, such as LWWIK, might cross the membrane more easily than others such as LWYIR because the energy barrier is lower.

The best structure and position of each peptide in the membrane were used to test the interaction with cholesterol as described in Materials and Methods. The final angular dynamics optimization of the procedure gave a series of complexes of low intramolecular and/or medium restraint energies. Comparative analyses of the noncovalent bonding occurring between the family of peptides and cholesterol show that the two parts of cholesterol (the fused ring structure and the hydroxyl) can be involved in the interaction. The

hydroxyl can be implicated in two interactions, one as a H-donor and a second as a H-acceptor. The ring can be involved in van der Waals contacts. We find that there is a unique feature of the interaction of LWYIK and cholesterol that is not seen with any of the other peptides of this series (Figure 8): the cholesterol hydroxyl can be the H-donor of the lysine terminal C=O group and the H-acceptor of the OH group of the tyrosine side chain. This fact is unique since in other complexes, the cholesterol hydroxyl is either a H-donor or a H-acceptor but not both, and sometimes, it is not involved in the interaction as illustrated for LWWIK (Figure 9). The interaction between cholesterol and LWYIK optimally satisfies the cholesterol polar head requirement, and it leaves the cholesterol rings rather free. With other peptides, lowering the energy of interaction results in the two molecules having significant van der Waals interactions, which occurs for the LWWIK–cholesterol complex (Figure 9), where the W indole ring lies parallel to the planar face of cholesterol.

**Mutagenesis of the Membrane-Proximal External Region (MPER) of HIV-1 gp41.** Our goal was to examine the effects of mutations in the LWYIK pentapeptide in the context of the entire HIV-1 Env. This peptide is part of the HIV-1 MPER in the gp41 ectodomain whose crucial role in viral fusion has been established (8). We employed the QuikChange mutagenesis method to substitute leucine (L) at position 679 with isoleucine (I). The mutation was introduced in the KpnI–BamHI NL4.3 envelope fragment that had been subcloned into vector pSP72. The mutated envelope fragment was then subcloned into envelope expression vector pSRHS. All mutations were confirmed by sequencing.

**Effects on Cell–Cell Fusion.** Cos-1 cells were transfected with WT and mutant pSRHS vectors and 48 h later were mixed with JC53-BL cells. Introduction of Tat through fusion results in the expression of firefly luciferase and  $\beta$ -galactosidase under transcriptional control of a minimal HIV LTR. The luciferase-based fusion assay showed that the L679I mutant fused cells 27% less efficiently than the wild-type construct (WT) (Figure 10).  $\beta$ -Galactosidase assays were performed by counting the number of blue syncytia per well and the average number of nuclei per syncytium. The latter provides a more reproducible measure of Env function since large syncytium formation requires multiple rounds of fusion (8). The average number of syncytia per well was reduced by 34% for the L679I mutant compared to WT, while the average number of nuclei per syncytium was reduced further to 48% of that of WT (Figure 10).

## DISCUSSION

The requirements for the CRAC motif are so flexible that there are many possible combinations for testing for their ability to sequester cholesterol. According to the CRAC algorithm, a CRAC motif can vary in length anywhere from 5 to 13 amino acids. For the longer peptides, there are 10 positions that can be any one of the 20 naturally occurring amino acids. This feature alone can generate  $1.024 \times 10^{13}$  possible sequences, and if one includes the variability in length, the number becomes much larger. It is unlikely that such a vast number of possible sequences can all be equally efficient to sequester cholesterol. In fact, it has been pointed out that the CRAC motif occurs very frequently in a few

Table 3: W Emission Properties

| peptide | methanol       |                    | buffer         |                    | SOPC           |                    | SOPC/cholesterol (1/1) |                    | ratio <sup>a</sup> |
|---------|----------------|--------------------|----------------|--------------------|----------------|--------------------|------------------------|--------------------|--------------------|
|         | $\lambda$ (nm) | relative intensity | $\lambda$ (nm) | relative intensity | $\lambda$ (nm) | relative intensity | $\lambda$ (nm)         | relative intensity |                    |
| LWYIK   | 336            | 1.7                | 341            | 0.74               | 340            | 0.71               | 340                    | 0.43               | 1.7                |
| LWFIK   | 336            | 1.3                | 341            | 0.75               | 340            | 0.61               | 340                    | 0.43               | 1.4                |
| LWYIR   | 335            | 2.0                | 340            | 1.0                | 340            | 0.73               | 340                    | 0.69               | 1.1                |
| LWYIH   | 335            | 1.6                | 341            | 0.91               | 342            | 0.82               | 342                    | 0.60               | 1.4                |
| LWWIK   | 336            | 1.95               | 340            | 0.95               | 339            | 0.90               | 341                    | 0.75               | 1.2                |
| LWLIK   | 335            | 2.4                | 340            | 1.3                | 340            | 1.0                | 340                    | 0.87               | 1.1                |
| IWYIK   | 336            | 2.0                | 340            | 0.87               | 339            | 0.67               | 339                    | 0.66               | 1.0                |

<sup>a</sup> The ratio is the ratio of the maximal emission intensity of the peptide in SOPC divided by the emission intensity of the peptide in SOPC and cholesterol (1/1).

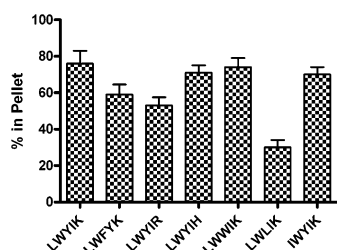


FIGURE 5: Partitioning of peptides into MLVs of SOPC and cholesterol (1/1) containing 10 mol % peptide. Error bars are the standard error of the mean of experiments repeated two times.

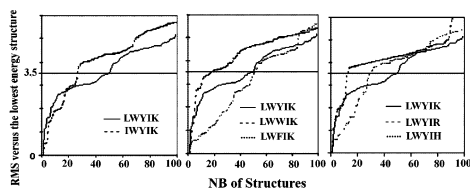


FIGURE 6: Analysis of peptide conformation diversity. For each peptide, 7776 conformations were calculated twice, once in a hydrophobic medium and once in water, under each condition, and the 50 best conformations were saved. Diversity of these structures was analyzed by calculating the all atom rms deviation with reference to the lowest-energy conformation. Structures were ranked by increasing rmsd. For the sake of analysis, a critical value, arbitrarily set at 3.5 rmsd, was used to analyze structural rigidity: the lower the number of conformations at that 3.5 rmsd value, the more rigid the structure.

genomes (30). In this work, we have focused on a specific pentapeptide CRAC motif, LWYIK, that has been shown to sequester cholesterol and is present in a protein, the gp41 protein of HIV, which associates with cholesterol-rich domains. We have blocked the terminal amino and carboxyl groups as they would be blocked in the intact protein. In this group of LWYIK analogues, we have not varied the W or I residues in the second and fourth position since these positions can be varied according to the CRAC algorithm, -L/V-(X)(1-5)-Y-(X)(1-5)-R/K-; i.e., they are the "X". We have, however, made peptides in which the first, third, and fifth residues of LWYIK are substituted. These three positions for a pentapeptide are required by the CRAC motif to be L/V, Y, and K/R, respectively.

The effects of the peptide on the enthalpy of the phospholipid chain melting transition in the presence of cholesterol (Table 1) provide a good criterion for identifying peptides that deplete a region of the membrane from cholesterol, resulting in an increase in the cooperativity and enthalpy of this transition. The suggestion that cholesterol is redistributed in the membrane and is sequestered in regions where it surpasses its solubility limit is supported by the

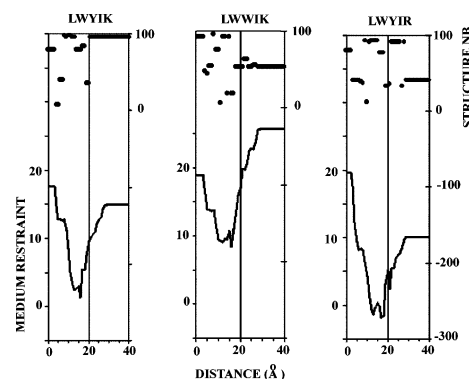


FIGURE 7: Restraint profile of insertion of LWYIK, LWWIK, and LWYIR into the membrane. The 100 conformations saved from stereosheet were independently tested for a systematic screening of insertion into a DPPC membrane. Then, a unique insertion profile was made by choosing at each membrane level the conformation that gave the minimal insertion restraint (bottom curve). The membrane center is at the  $x$  axis origin. At 20 Å is the membrane–water surface. From 0 to 15 Å is the acyl chain layer. From 15 to 20 Å is the membrane–water interface (lipid polar headgroups). At >20 Å is the water–medium interface. At each level, the peptide is located by a point at its mass center. The rank number of the conformations giving the best insertion is plotted on the top of the figure to demonstrate that optimal insertion involves different conformations of peptides. The restraint values are calculated as described in ref 28 as the sum of two restraints, one simulating the bilayer hydrophobicity ( $E_{pho}$ ) and the other the lipid perturbation ( $E_{lip}$ ). The lowest restraint values correspond to the best stability. Peptides with a lower restraint value in the membrane than in water should be able to cross the membrane. Peptides with a lower restraint value in the lipid polar head interface than in the membrane core will more easily remain at the interface, rather than cross the membrane.

observation that anhydrous cholesterol crystallites are formed (Table 2). Among the peptides used in this study, *N*-acetyl-LWYIK-amide is the most potent in increasing the enthalpy of the SOPC transition and appears to be the only one to do so at 30% cholesterol (Figure 4). With a higher cholesterol concentration, LWYIK has the strongest cholesterol-sequestering effect, followed by *N*-acetyl-LWFIK-amide and then *N*-acetyl-LWYIR-amide and *N*-acetyl-LWYIH-amide that have comparable potencies (Table 1). *N*-Acetyl-LWWIK-amide and *N*-acetyl-LWLIK-amide have still weaker potency, and *N*-acetyl-IWYIK-amide is the poorest in forming regions in the membrane devoid of cholesterol.

Differences in membrane partitioning among the peptides (Figure 5) should not account for their different effects on the phase transitions of SOPC/cholesterol mixtures. Indeed, all peptides prefer to be at the membrane interface. Interestingly, *N*-acetyl-IWYIK-amide has the strongest effect in lowering the enthalpy of the transition for the pure SOPC,



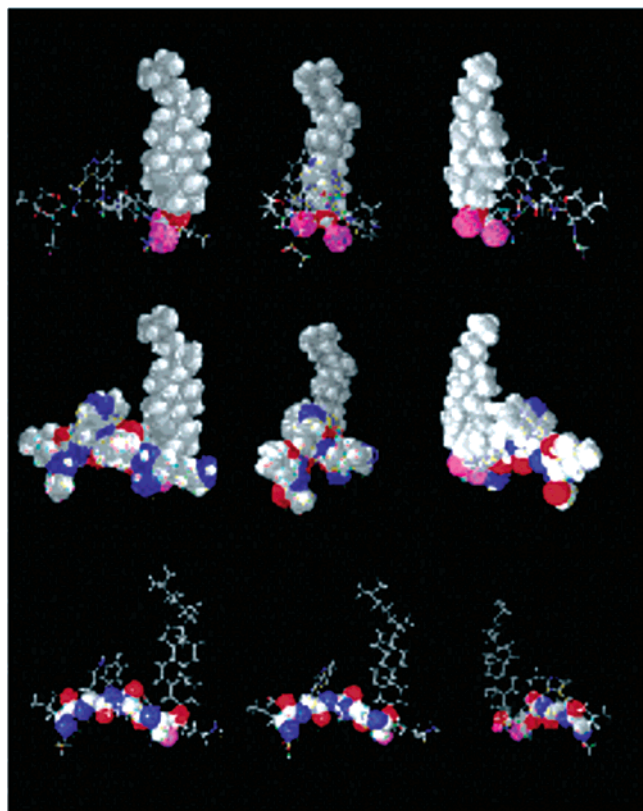


FIGURE 8: Illustration of the lowest-energy complex between cholesterol and LWYIK in the membrane. Positions of the two molecules were systematically tested and minimized by angular dynamics (32). The lower-energy complex for intra- and intermolecule interactions is here shown in three different positions (A–C) with three different outlooks. (1) Cholesterol is drawn in CPK, and the peptide bonds are sticks. (2) All atoms (cholesterol and peptide) are in CPK. (3) Backbone atoms (N–C–CO) are drawn in CPK, and all other bonds are sticks. In all cases, the oxygens of the tyrosine OH group and of the lysine backbone CO group are colored pink.

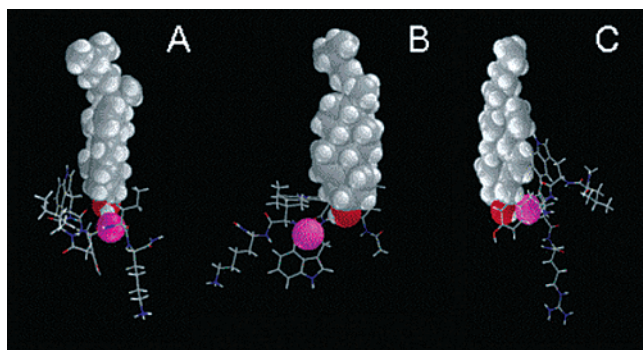


FIGURE 9: Illustration of the lowest-energy complexes between cholesterol and IWYIK (A), LWYIK (B), and LWYIR (C), and cholesterol in the membrane. The lower-energy molecules for intra- and intermolecule interactions are shown here with a CPK image of the cholesterol atoms, and the closest peptide oxygen in pink-colored CPK (in each case, it was the third residue backbone C=O group).

indicating that the weaker effect of this peptide on cholesterol-containing membranes is not a consequence of its inability to partition into membranes.

There is also little difference in the maximum emission wavelength among the seven peptides in the presence of lipids (Table 3). This indicates that the differences in the extent of sequestering of cholesterol observed by DSC are

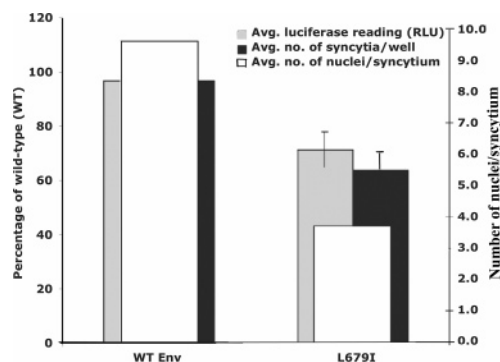


FIGURE 10: Cell–cell fusion assays of the wild type and MPER mutant L679I. Data are the average of three runs, for each of which samples were run in duplicate. The number of syncytia per well was determined by counting the syncytia in each well (24-well plate). The number of nuclei per syncytium was determined by counting 20 syncytia per well. Luciferase readings (RLU, relative light units) were taken as per the manufacturer's instructions.

not a consequence of a difference in hydrophobicity of the peptides that would result in a change in the depth of insertion of the peptide into the membrane. However, although the emission wavelength does not change, in some cases the intensity of emission does change. The peptides with Y can undergo resonance energy transfer to W. *N*-Acetyl-LWWIK-amide has the lowest emission intensity per W of all of the peptides (given that it has two W residues). There can be resonance energy transfer between the two W residues, but this will not result in a change in emission intensity, if both W residues emit at the same wavelength and have the same quantum yields. The peptide *N*-acetyl-LWLIK-amide has the highest emission intensity, even though this peptide cannot exhibit resonance energy transfer. This is true even with solutions of the peptide in methanol or in buffer. The fact that the quantum yields of these peptides vary indicates that even though these peptides are small and likely very flexible, there is sufficient difference in their preferred folding that results in differences in the environment of the W residue.

In general, the fluorescence emission intensity from the W residue of these peptides is lower in the presence of membranes containing cholesterol than it is in pure SOPC vesicles (Table 3). Interestingly, the ratio of fluorescence emission intensity of each peptide in the presence of SOPC versus SOPC and cholesterol (1/1) is generally related to the ability of the peptide to sequester cholesterol into domains. Thus, *N*-acetyl-LWYIK-amide, which is the most potent in sequestering cholesterol, exhibits the largest difference in W emission intensity between membranes with or without cholesterol. Conversely, the W emission of the peptide *N*-acetyl-IWYIK-amide, which shows no tendency to preferentially partition into cholesterol-rich domains, is unaffected by cholesterol (Table 3). We suggest that *N*-acetyl-IWYIK-amide is sequestered into cholesterol-depleted domains, and therefore, its fluorescence is not affected by the presence of cholesterol. The DSC results support this suggestion because *N*-acetyl-IWYIK-amide lowers the transition enthalpy of pure SOPC as well as that of SOPC in mixtures with cholesterol. The fluorescence study provides another independent parameter, in addition to the DSC results, to show which peptides are affected by cholesterol in the membrane.



The peptide that causes the greatest rearrangement of cholesterol in membranes is *N*-acetyl-LWYIK-amide. This peptide has a CRAC motif. Major factors contributing to the differences among the peptides used in this work include their structural flexibility, their position at the interface, and their noncovalent bonding with cholesterol. A comparison of the cholesterol sequestering action of these analogue peptides provides good support for the concept of the CRAC motif. The peptide *N*-acetyl-LWFIK-amide is not formally a CRAC peptide, and it is not as selective as *N*-acetyl-LWYIK-amide in sequestering cholesterol at low concentrations. *N*-Acetyl-LWFIK-amide differs from *N*-acetyl-LWYIK-amide by only a single OH group. This OH group is involved as a H-donor in the LWYIK-cholesterol complex. The H-bond interaction is with the C=O group of the terminal lysine of LWYIK which acts as a H-acceptor in the complex. *N*-Acetyl-LWWIK-amide has much lower cholesterol-sequestering activity, and it is not a CRAC motif. In addition, the weakest peptide, *N*-acetyl-IWYIK-amide, has the same tyrosine groups as the best cholesterol-sequestering peptide, *N*-acetyl-LWYIK-amide. The IWYIK peptide does not correspond to a CRAC motif. In this latter case, the reduced conformational flexibility of the peptide may inhibit the formation of stable interactions with cholesterol. Thus, aromatic groups are likely important for stacking with cholesterol, but the conformation of the peptide and its presentation to the membrane are also important in determining its ability to preferentially bind to cholesterol-rich regions of the membrane. The central Y residue in LWYIK may have a particular role in stabilizing interactions with cholesterol by forming H-bonds (Figure 10). However, as expected because of the short length and flexibility of these peptides, this feature is necessary for cholesterol selectivity but is not sufficient. Changing K to R weakens the capacity to interact with cholesterol because the polar head of R tends to force the peptide out toward the water interface, and in addition, it decreases the motif's structural flexibility.

On the basis of NMR evidence, it was suggested that one of the aromatic groups of *N*-acetyl-LWYIK-amide stacks with the A ring of cholesterol (7). The results of modeling support the idea that, under the best energy conditions, the LWYIK motif is located within the interface wrapped around the polar head of cholesterol. However, under less favorable conditions, more frequent with other peptides, aromatic groups stack with the A ring of cholesterol. More in concordance with modeling is the NMR and mutational evidence showing that the Y residues of the benzodiazepine receptor are important for the interaction with cholesterol (31).

We further tested the IWYIK segment in the context of an intact gp41 molecule. The L679I mutant of gp41 had significantly less fusogenic activity than the wild-type protein (Figure 9). This mutation did not completely eliminate the ability of gp41 to induce formation of syncytia, but it did lower the activity despite the fact that the substituted amino acid, isoleucine, has properties similar to those of the native residue, leucine. Further studies are currently underway to determine the cause for the lower activity of the L679I mutant. From the results of the peptide studies, it seems likely that the decreased infectivity of this mutant is a result of its attenuated interaction with raft domains.

This is the first study to systematically investigate the properties of a family of closely related peptides in terms of their ability to promote rearrangement of cholesterol in membranes. It is found that their effectiveness is very dependent on small changes in their amino acid composition. It can result in the formation of cholesterol-rich domains either by being excluded from cholesterol-rich domains, as is the case for *N*-acetyl-IWYIK-amide, or by preferentially binding to these domains, which occurs with *N*-acetyl-LWYIK-amide. There is some relationship to the rules for a CRAC motif in that peptides that fulfill the requirements of this motif are often found to sequester cholesterol by binding to cholesterol-rich domains. However, this criterion is not exact. Furthermore, there may be differences between the properties of the isolated small peptide and the behavior of the same sequence when it is part of a larger protein, although in the example presented here, the L679I mutant of gp41 also exhibits decreased biological activity. It seems likely that no general rule will be able to predict with exactitude if a protein segment will recruit cholesterol in a membrane, although the presence of a CRAC motif adjacent to a transmembrane helix is suggestive of such an event taking place.

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